

Developments towards Machine-driven Design & Discovery of Organic Semiconductors

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7th Annual Commonwealth Computational Summit
on AI in Education, Medicine, and Research

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Shi Li (Graduate, PhD)
Josiah Roberts (Graduate, PhD)
E. Kirkbride Loya (Graduate)
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Keerthan Rao (Graduate, PhD)
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Jodie Canada (Undergraduate)
Camron De'vine (Undergraduate)

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Corrine Elliott (Undergraduate)
Tristan Finn (Undergraduate)
Kate Fraser (Undergraduate)
Stephen Goodlett (Undergraduate)
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Asare Nkansah (Undergraduate)
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Andrew Smith (Undergraduate)
William Allen Smith (Undergraduate)
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Hanna Suarez (Undergraduate)
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Award No. 1627428
Cooperative Agreement No. 2019574

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@cmrisko @riskolab



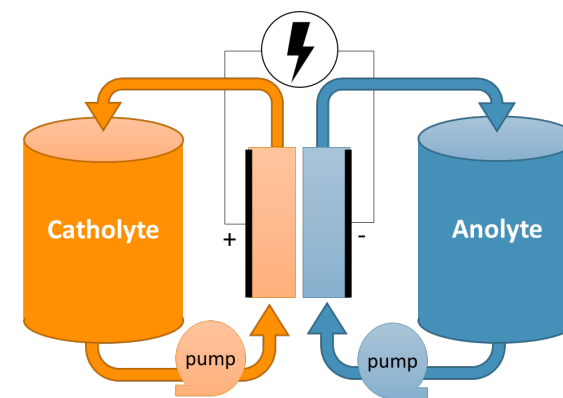
Organic materials for electronics, power generation, and storage



Samsung

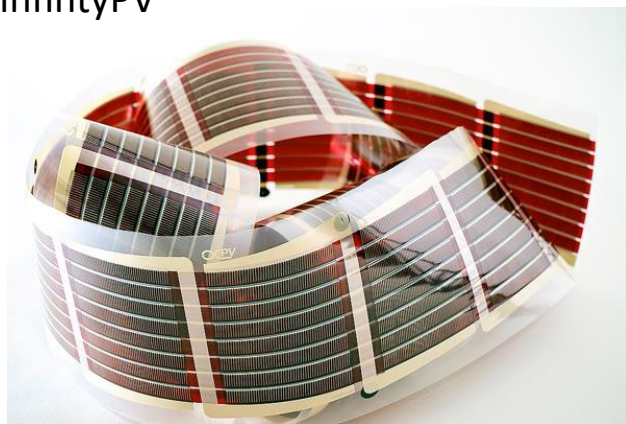


University of Tokyo

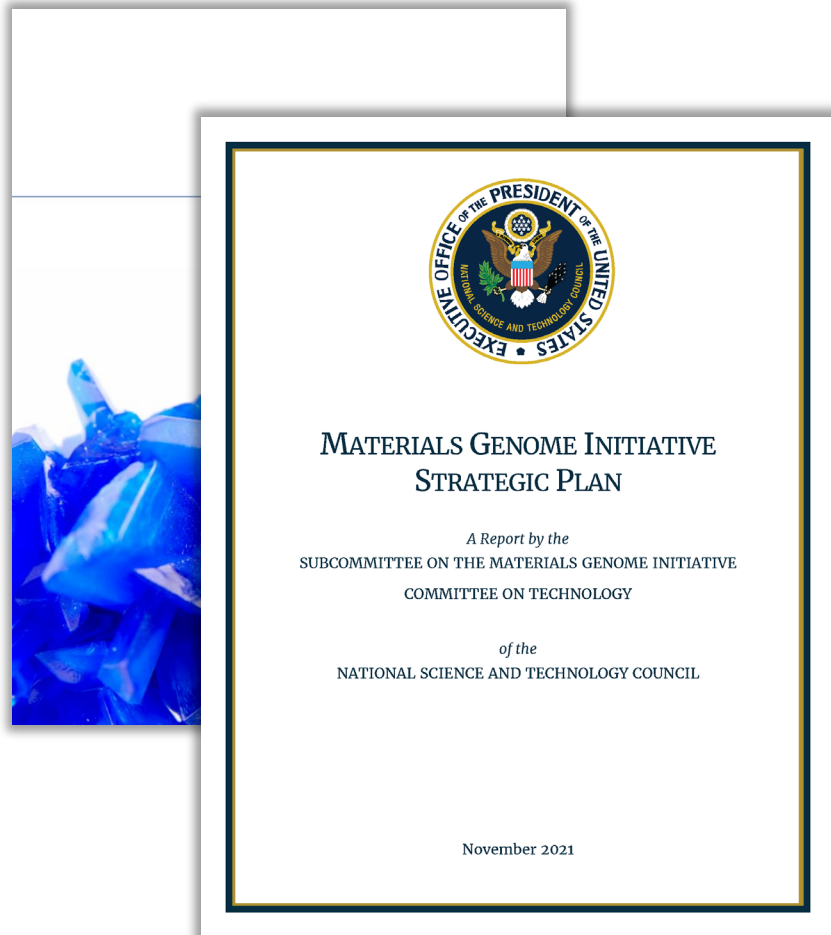


LG

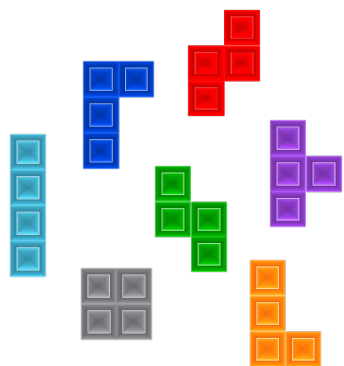
infintyPV



LG

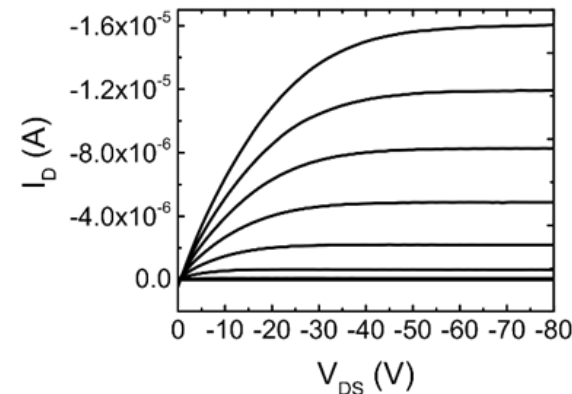


Launched in 2011 to accelerate the discovery, design, development, and deployment of new materials, at a fraction of the cost, by harnessing the power of data and computational tools in concert with experiment.



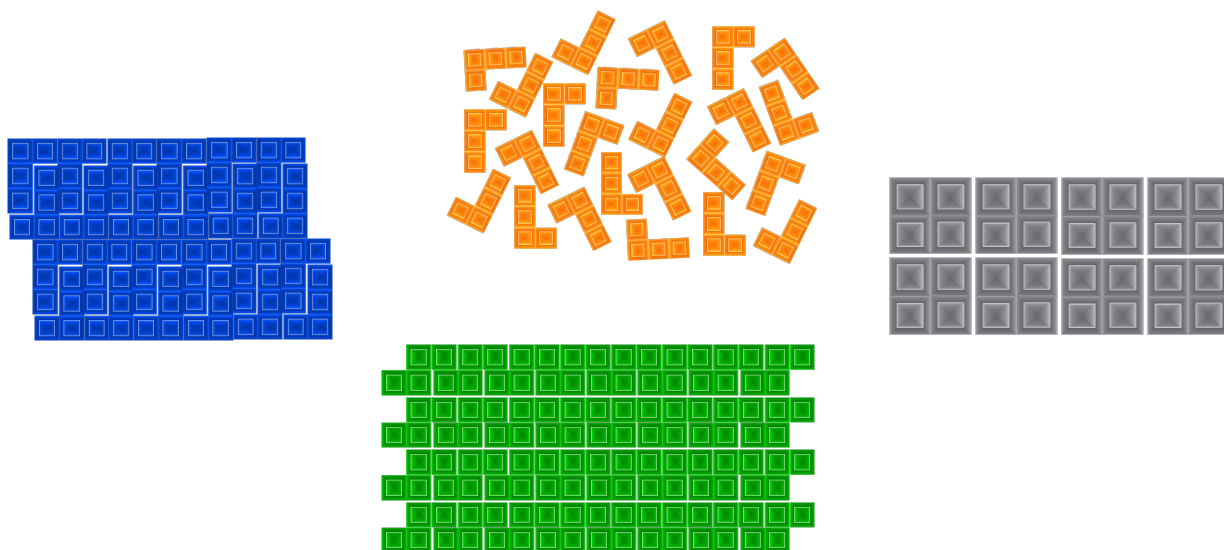
A vast chemical space of organic chromophores

BUILDING BLOCK



A. Zeidell *et al.* Chem. Mater. (2019) 31, 6962.

RESPONSE

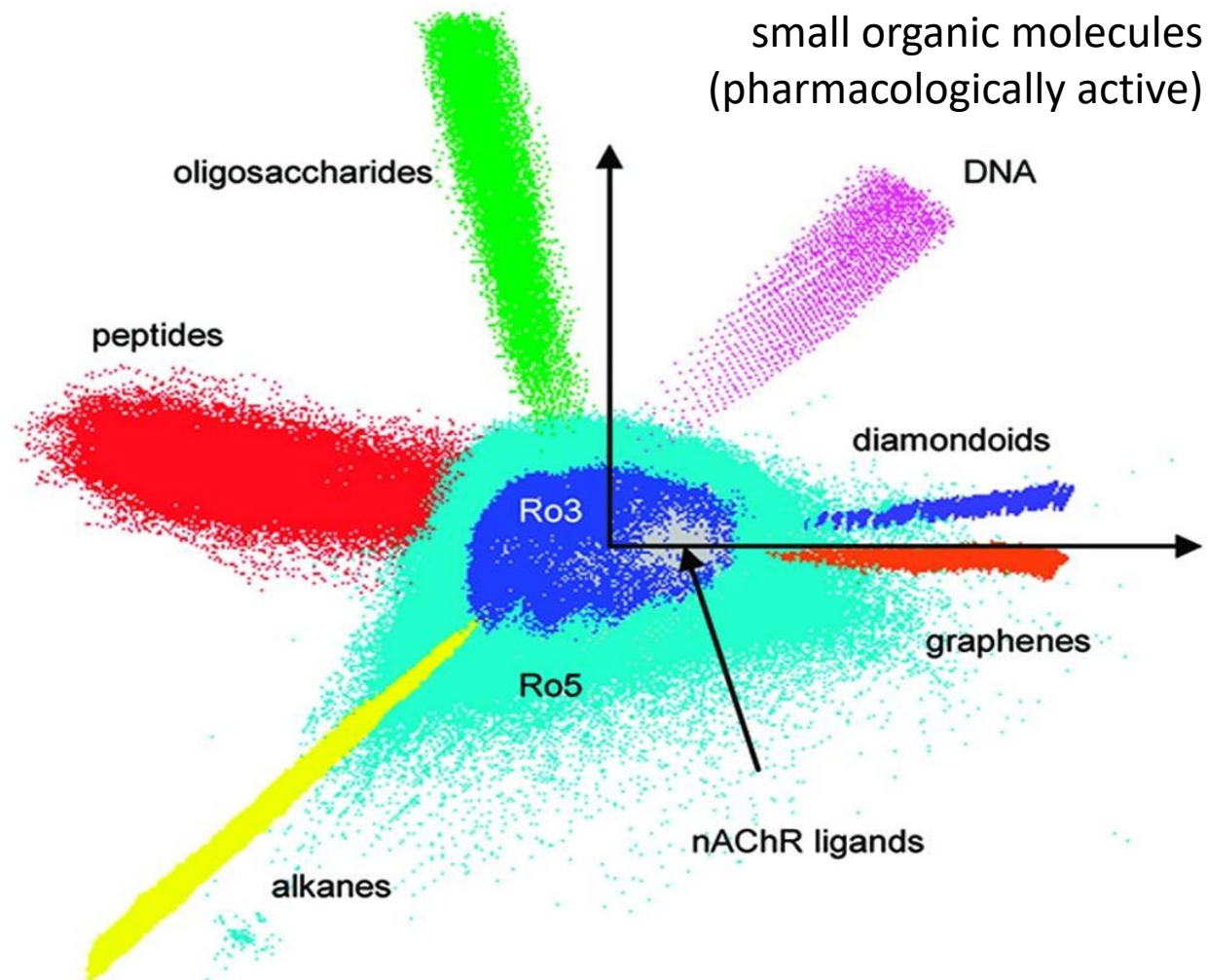


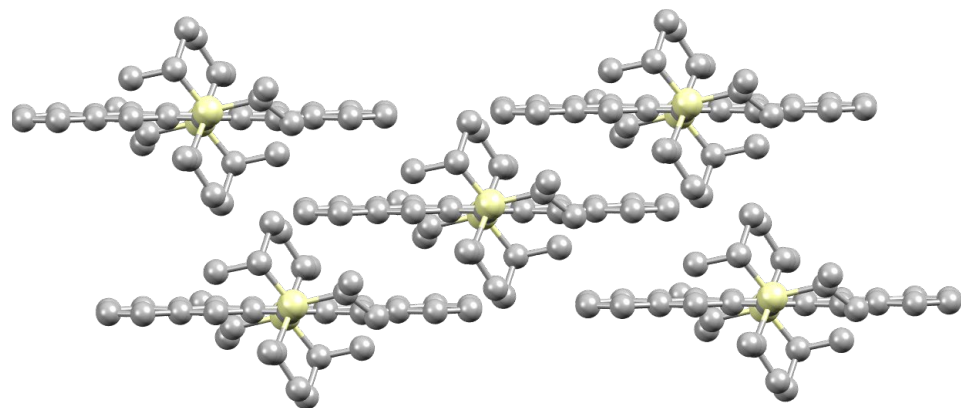
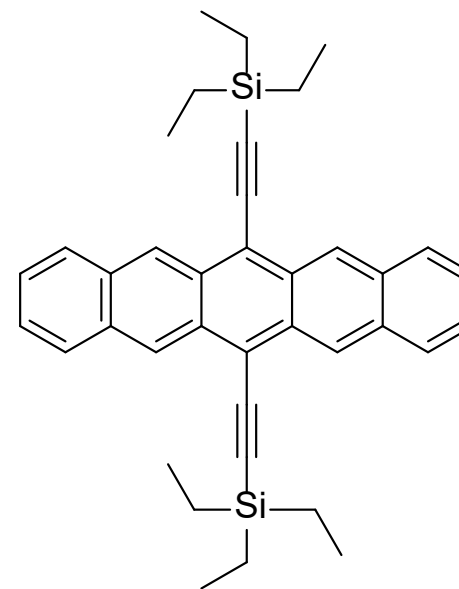
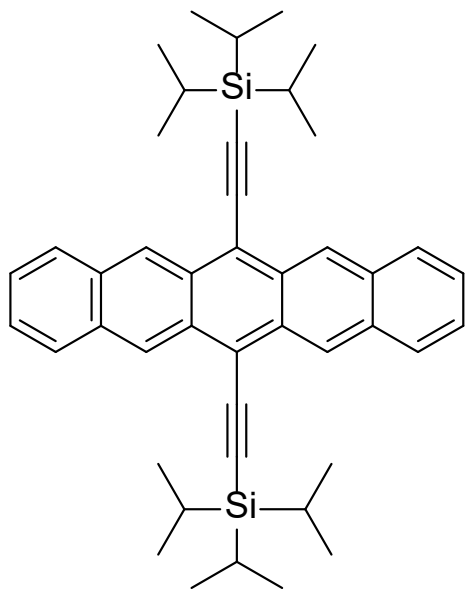
**STRUCTURE & SYNTHESIS
PREDICTION**

**PROPERTY
PREDICTION**

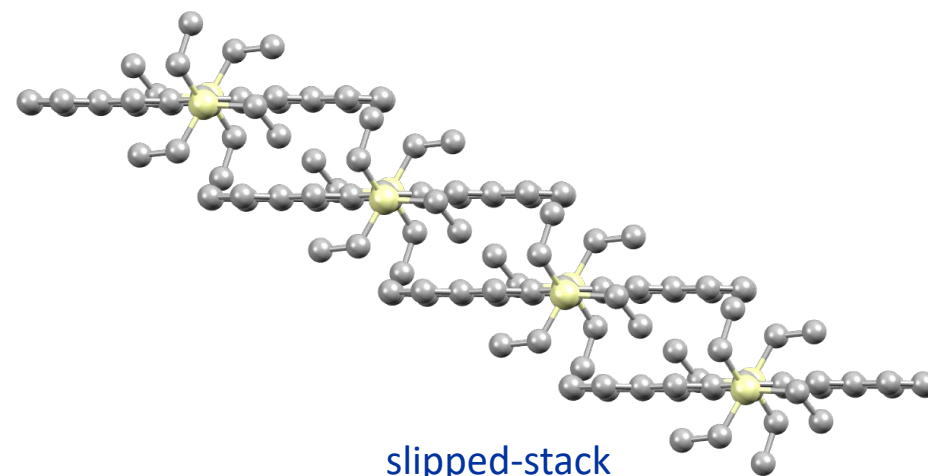
All things considered...chemical space is massive!

- 10^{18} grains of sand on Earth
- 10^{23} stars in the visible sky
- 10^{60} small organic molecules (pharmacologically active)



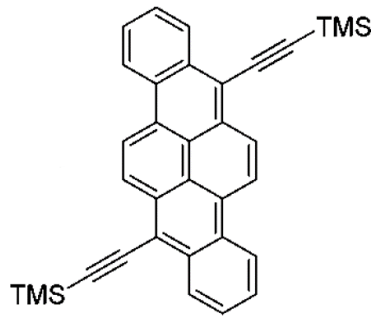


brickwork



slipped-stack

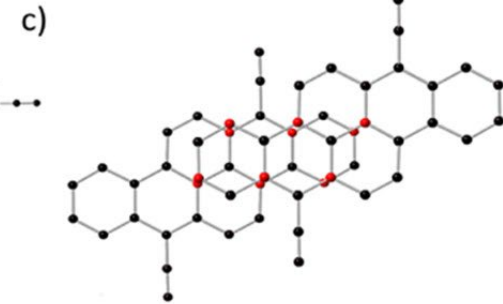
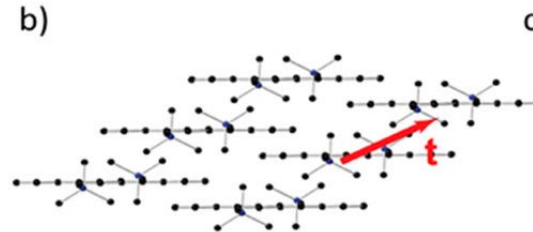
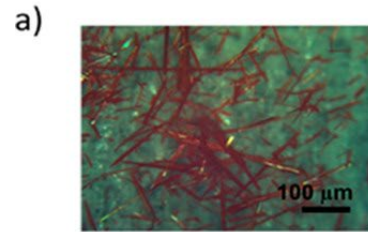
$$\mu_h = 0.028 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$$



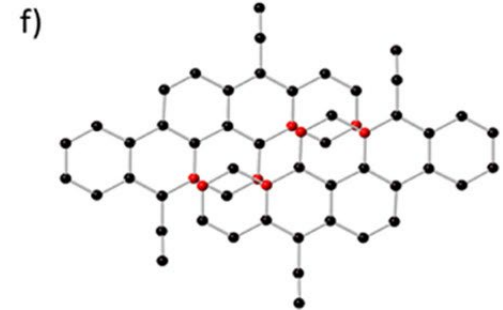
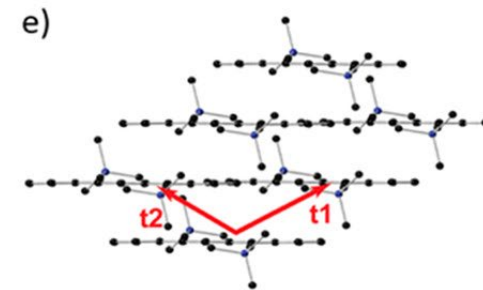
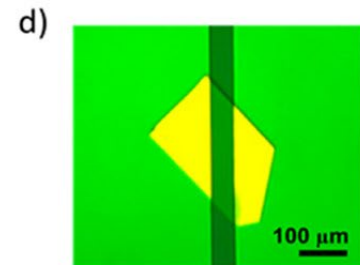
low T processing

high T processing

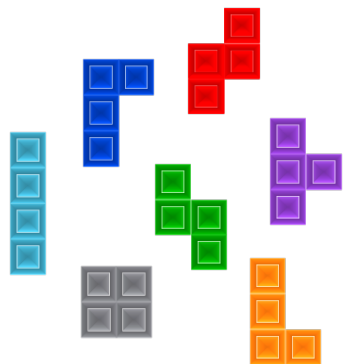
TMS-DBC LT (red needles)



TMS-DBC HT (yellow plates)

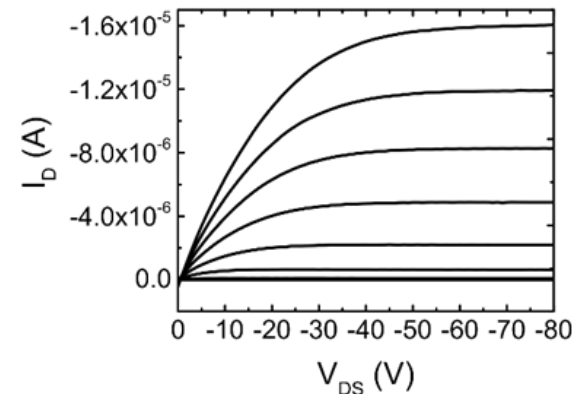


$$\mu_h = 2.1 \text{ cm}^2 \text{V}^{-1} \text{s}^{-1}$$



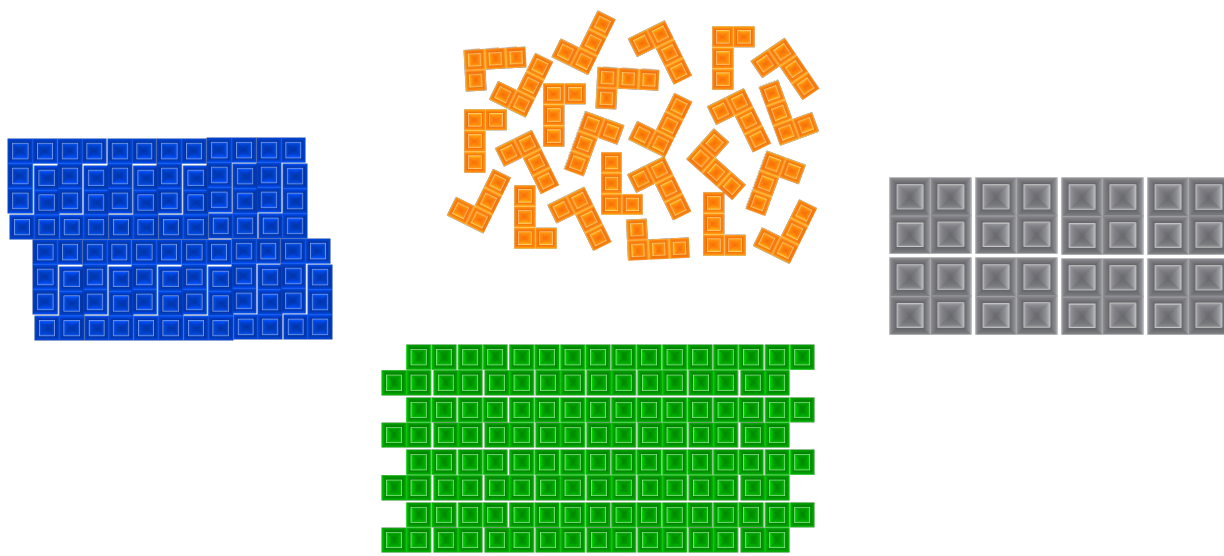
A vast chemical space of organic chromophores

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A. Zeidell *et al.* Chem. Mater. (2019) 31, 6962.

RESPONSE



**STRUCTURE & SYNTHESIS
PREDICTION**

**PROPERTY
PREDICTION**



Organic Crystals in Electronic and Light Oriented Technologies (OCELOT)

Open Access infrastructure



oscar.as.uky.edu



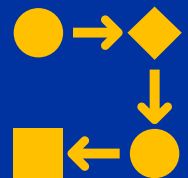
56k crystals

47k molecules

38k π -conjugated chromophores



A Python tool kit for data analysis



High-throughput computational workflows



User-friendly web user interface

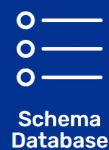


Access to machine learning predictions



OCELOT

OCELOT is an online archive for
Organic Crystals in Electronic and Light-Oriented Technologies



Database Statistics

as of March 26, 2023, 8:32 p.m. (US/Eastern)



56296

Crystals

with structure determined experimentally by XRD/ND.



2000

Descriptors

with optoelectronic properties computed via state-of-art DFT.



10e6

Core hours

have been used to fill this database.



1853

Researchers

are using this for their research since Feb 14, 2021.

Since 2021 launch →

- 2700+ users
 - 170 registered users that can access tools
- 48 countries

Tools →

- XtalTransform**

generate crystal structures based on templates

- FilmDiffract**

generate X-ray thin film diffraction patterns based on bulk crystal structures

- RepCheck**

reproducibility checker for experimental procedures

- OCELOT-ML**

open-access portal to ML models



Organic Crystals in Electronic and Light Oriented Technologies (OCELOT)

Chromophore

[IP-tuned LC- ω HPBE/Def2SVP]

Electronic structure - Ground State

Vertical IP	6.365 eV
Adiabatic IP	6.321 eV
Vertical EA	-1.932 eV
Adiabatic EA	-1.959 eV
Hole reorganization energy	0.092 eV
Electron reorganization energy	0.130 eV
HOMO-LUMO gap	4.466 eV
Tuned w	0.178

Electronic structure - Excited State

Singlet transition (S0->S1)	1.865 eV
Triplet transition(S0->T1)	0.163 eV
Adiabatic (S0->T1)	0.616 eV

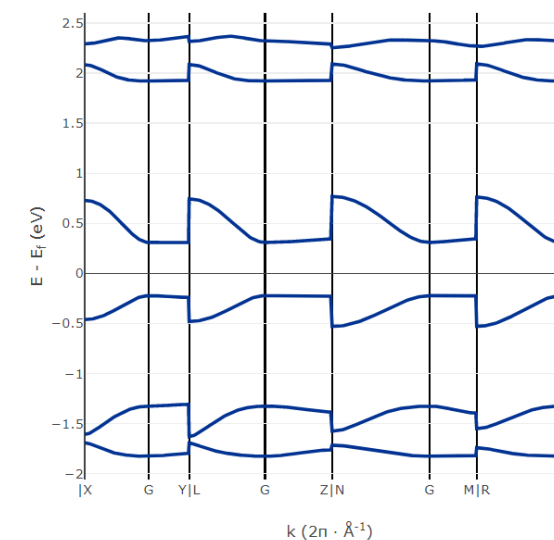
Crystal

[Periodic DFT | PBE with projector-augmented wave method]

Electronic Structure

Fermi energy	0.847 eV
Bandgap	0.532 eV
Bandgap is direct	False
Fitted bandgap	1.637 eV
Valence band maximum	0.623 eV
Hole line effective mass	0.889
Hole effective mass tensor	[1.189, 6.763, 314.853]
Valence band dispersion	0.239 eV
Conduction band minimum	1.155 eV
Electron line effective mass	0.480
Electron effective mass tensor	[83.278, 3.901, 0.624]
Conduction band dispersion	0.420 eV
Line segment	X-G:[0. -0.5 0.]-[0. 0. 0.]
K-point	[0.0, -0.067, 0.0]

Bandstructure



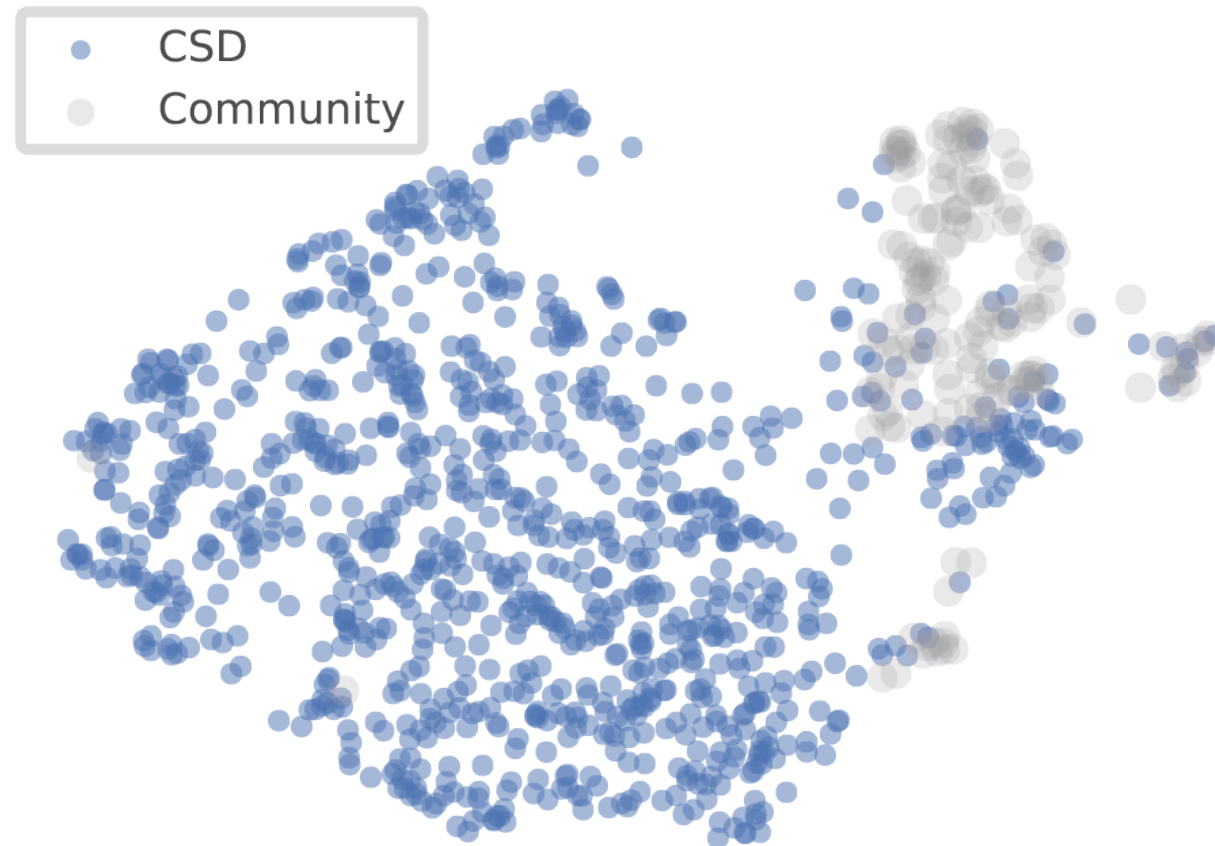
Percolation (DFT)

H-H coupling	0.059 eV
H-H dimer	Long axis slip: 6.29 Å Short axis slip: 0.93 Å Vertical distance: 3.39 Å
L-L coupling	0.113 eV
L-L dimer	Long axis slip: 6.29 Å Short axis slip: 0.93 Å Vertical distance: 3.39 Å

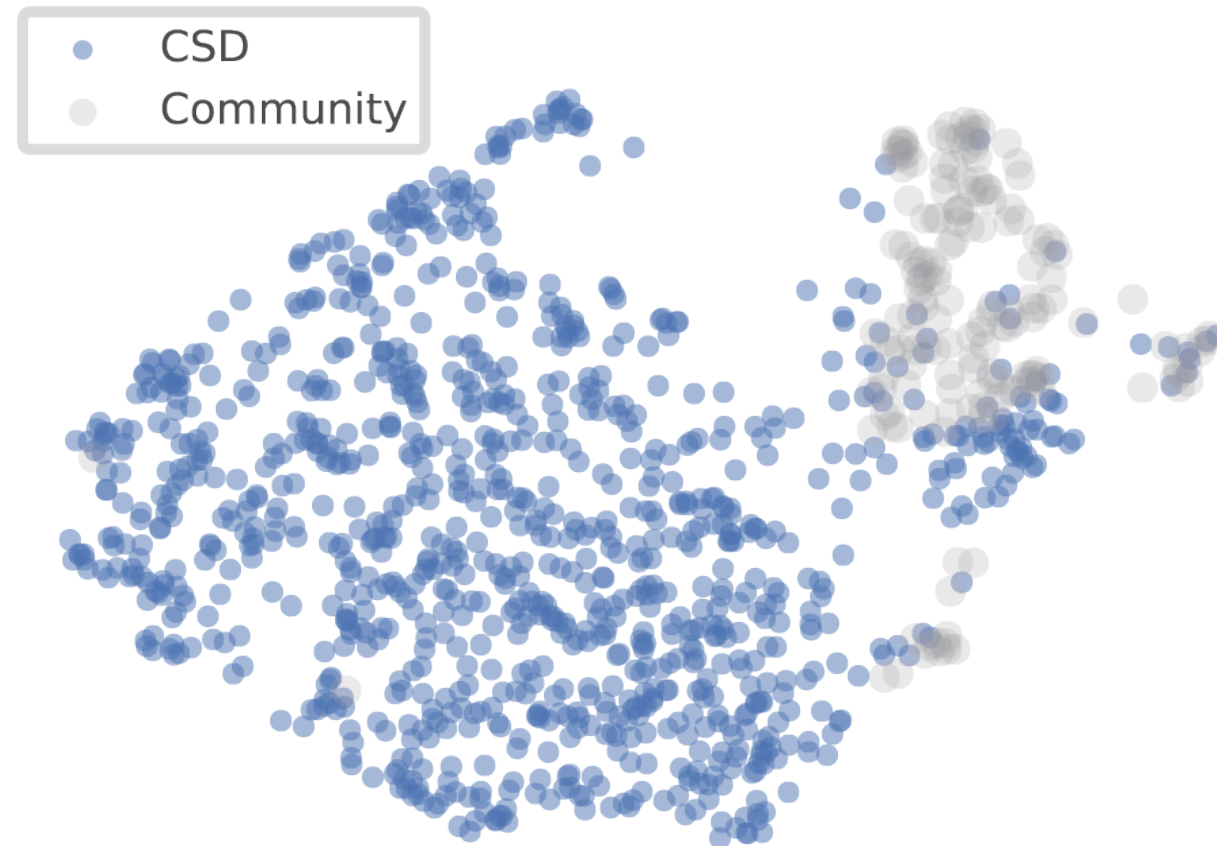
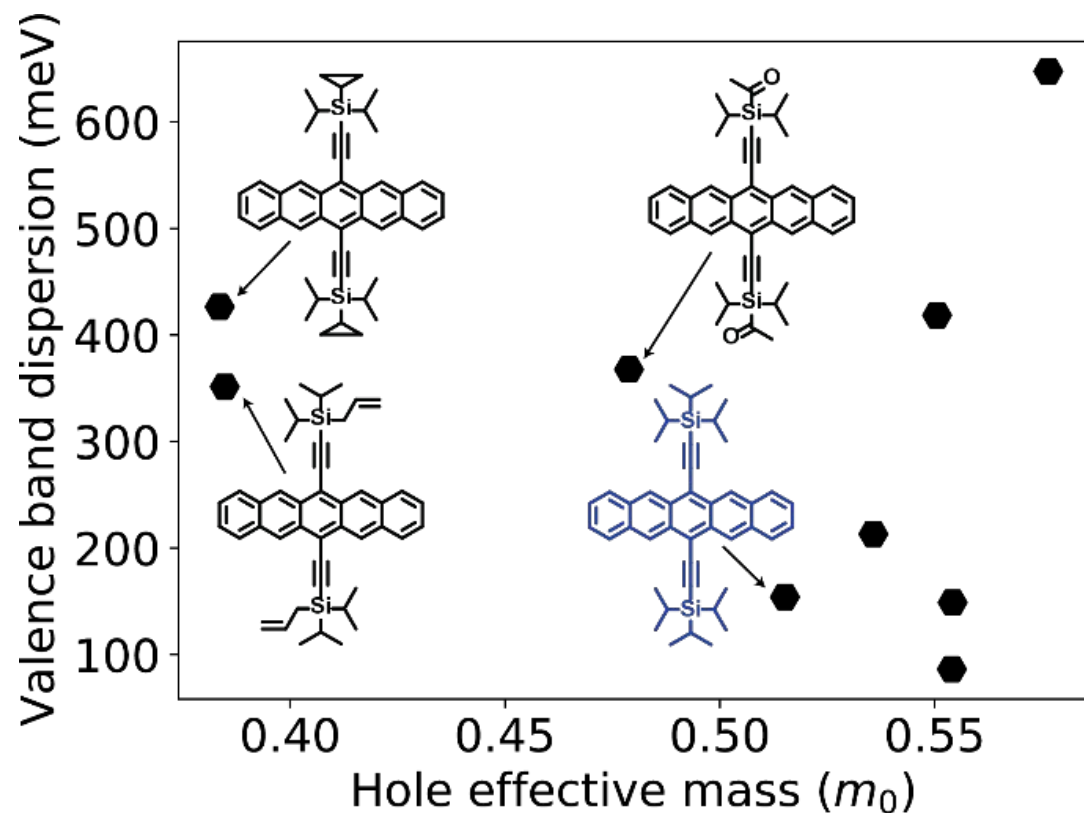
OCELOT Application Programming Interface (API)

https://github.io/ocelot_api

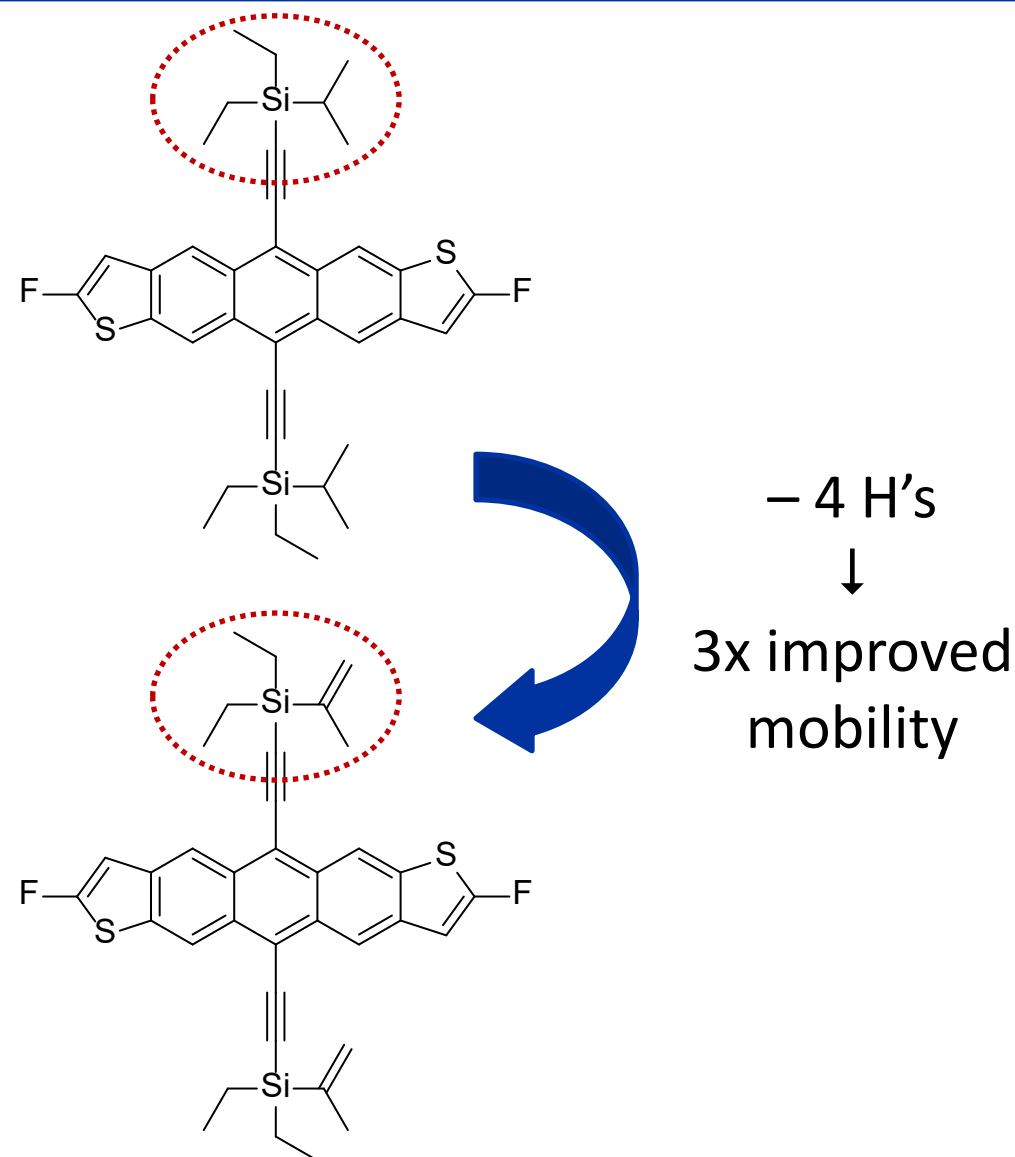
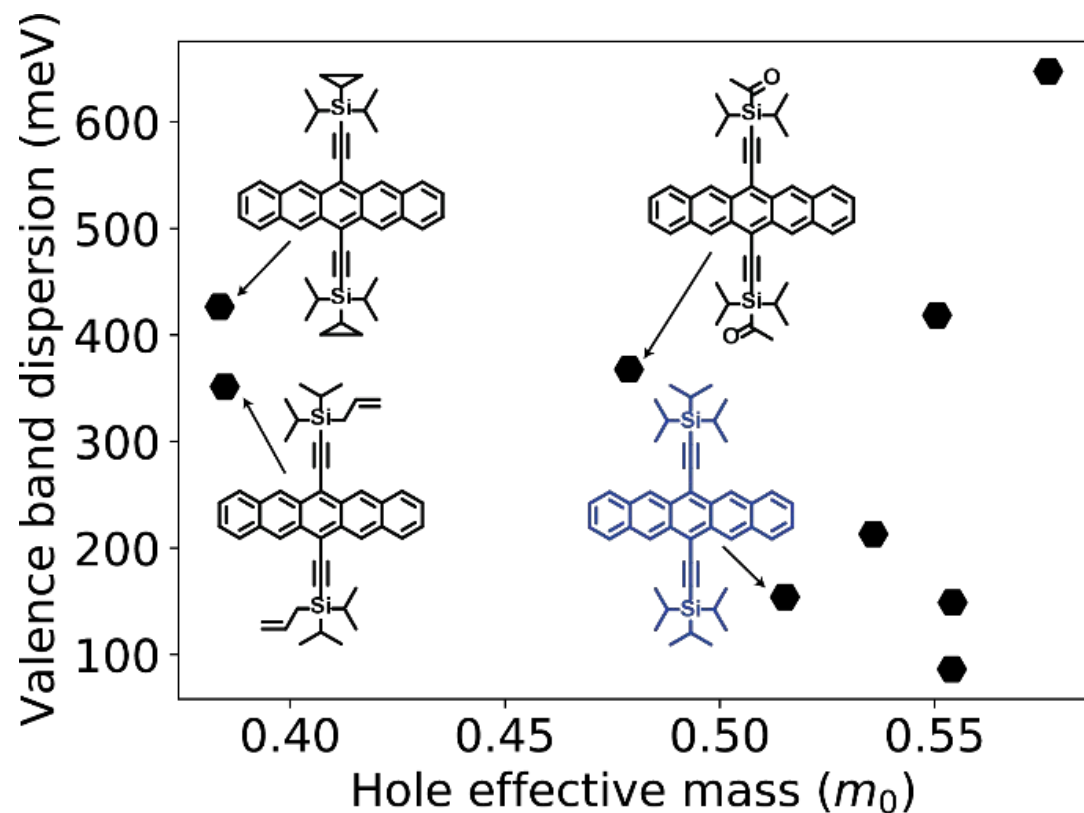
- OCELOT contains structures not reported elsewhere – ‘dark’ or missing structures from the literature

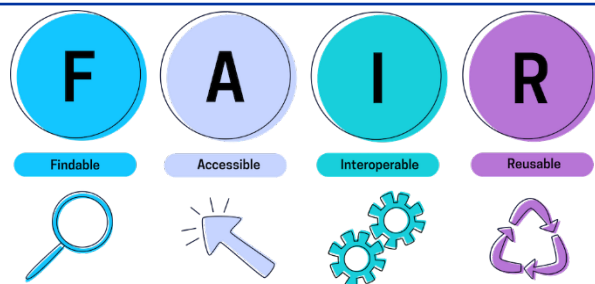


- Inclusion of dark structures aids in further exploration of chemical space of OSC



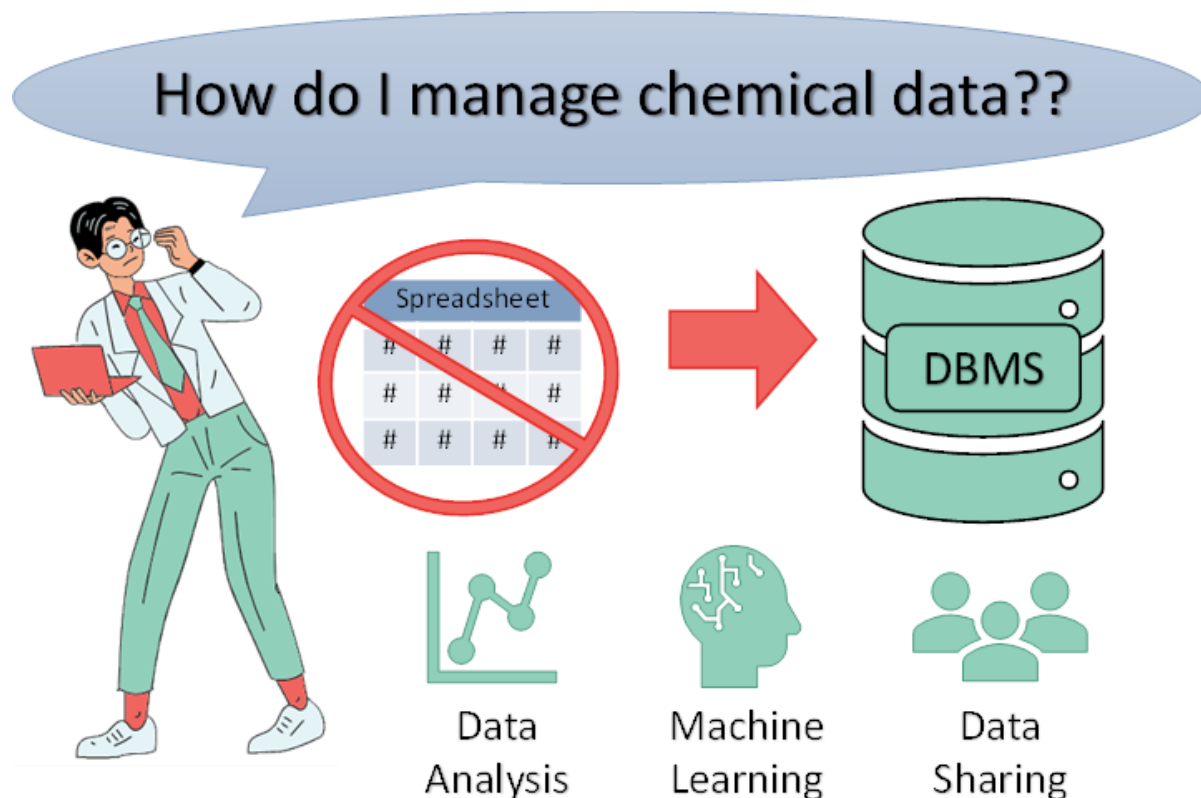
- Inclusion of dark structures aids in further exploration of chemical space of OSC



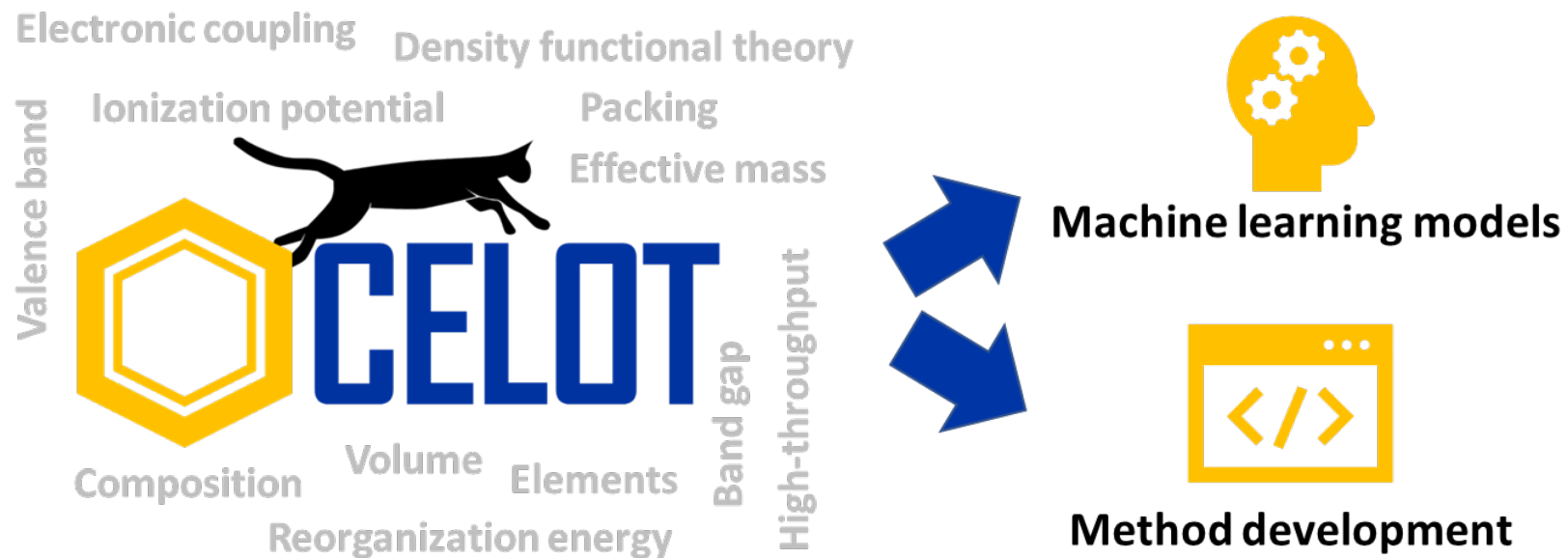


<https://flur.ee/why-fluree/>

- ❑ **Findable**
 - Could anyone find the data they are looking for within my data?
- ❑ **Accessible**
 - Could anyone access my data?
- ❑ **Interoperable**
 - Could anyone in my field understand the way my data is organized?
- ❑ **Reusable**
 - Could anyone use my data for validation or for their own analysis ?

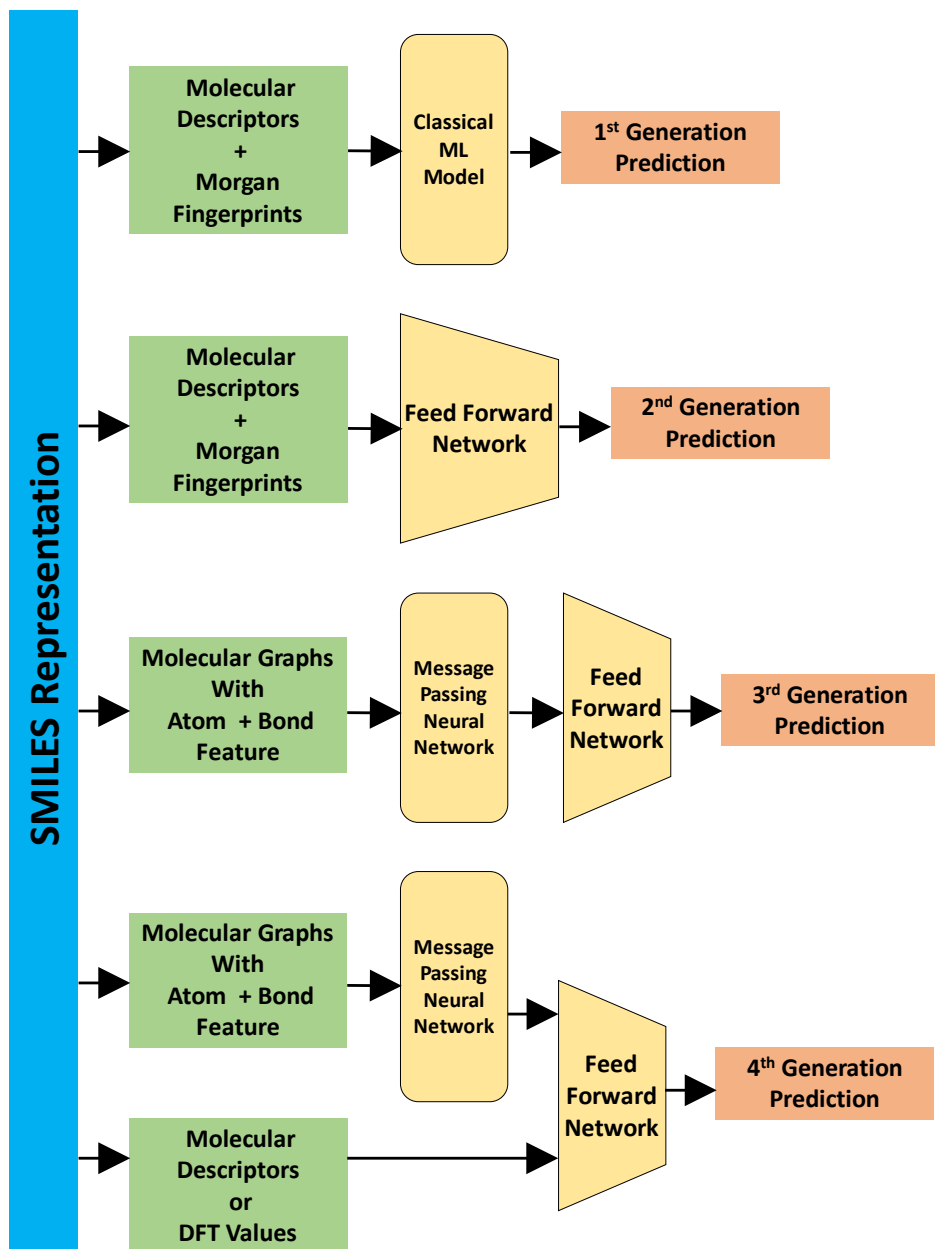


Let's predict molecular & materials properties



How can we use this molecular & materials data to predict properties of new systems?

Let's predict molecular & materials properties



- Challenges that need to be overcome:
 - Computing DFT-based electronic properties is time consuming
 - Recently published models are trained on datasets with ~25 atoms or rather niche chemical space
- Properties modeled:
 - Vertical (VIE) and adiabatic (AIE) ionization energies
 - Vertical (VEA) and adiabatic (AEA) electron affinities
 - Cation (CR) and anion (AR) relaxation energies
 - HOMO energies (HOMO), LUMO energies (LUMO)
 - HOMO-LUMO energy gaps (H-L)
 - Electron (ER) and hole (HR) reorganization energies
 - Lowest-lying singlet (S0S1) and triplet (S0T1) excitation energies



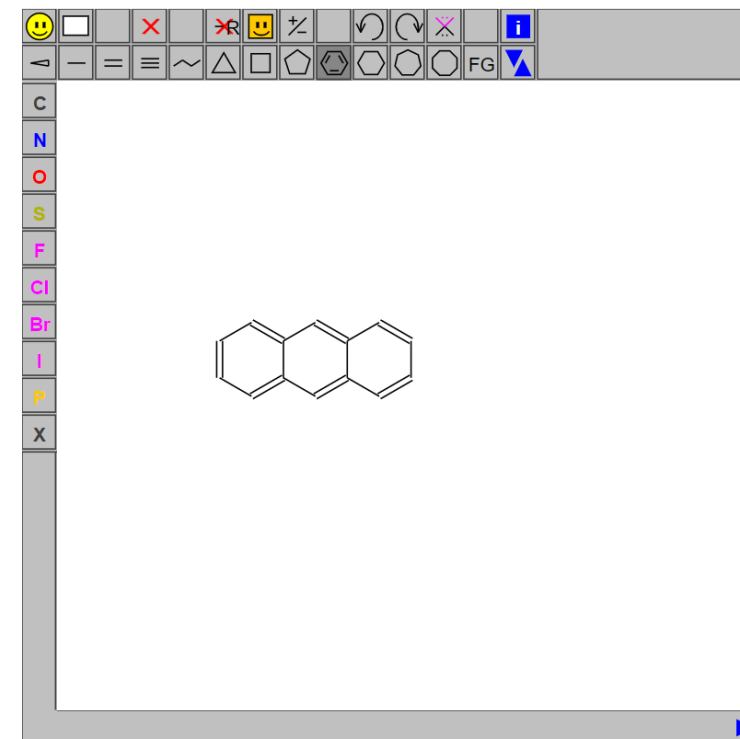
- OCELOT ML provides the infrastructure to make the models publicly accessible
- Currently, molecular predictions with SMILES input is supported
- Best models from our training are available



OCELOT ML



oscar.as.uky.edu/ocelotml



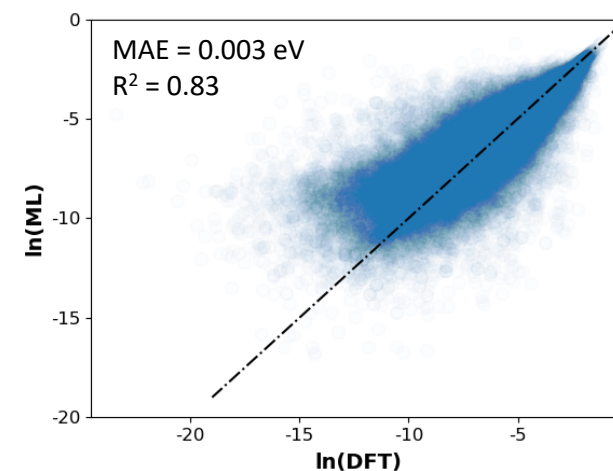
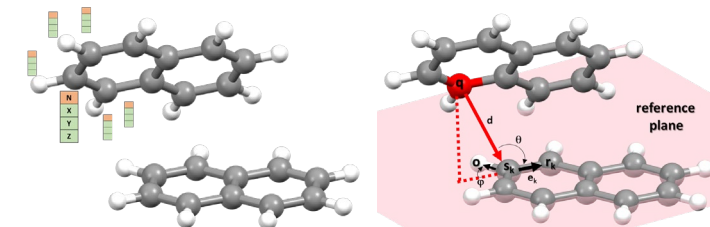
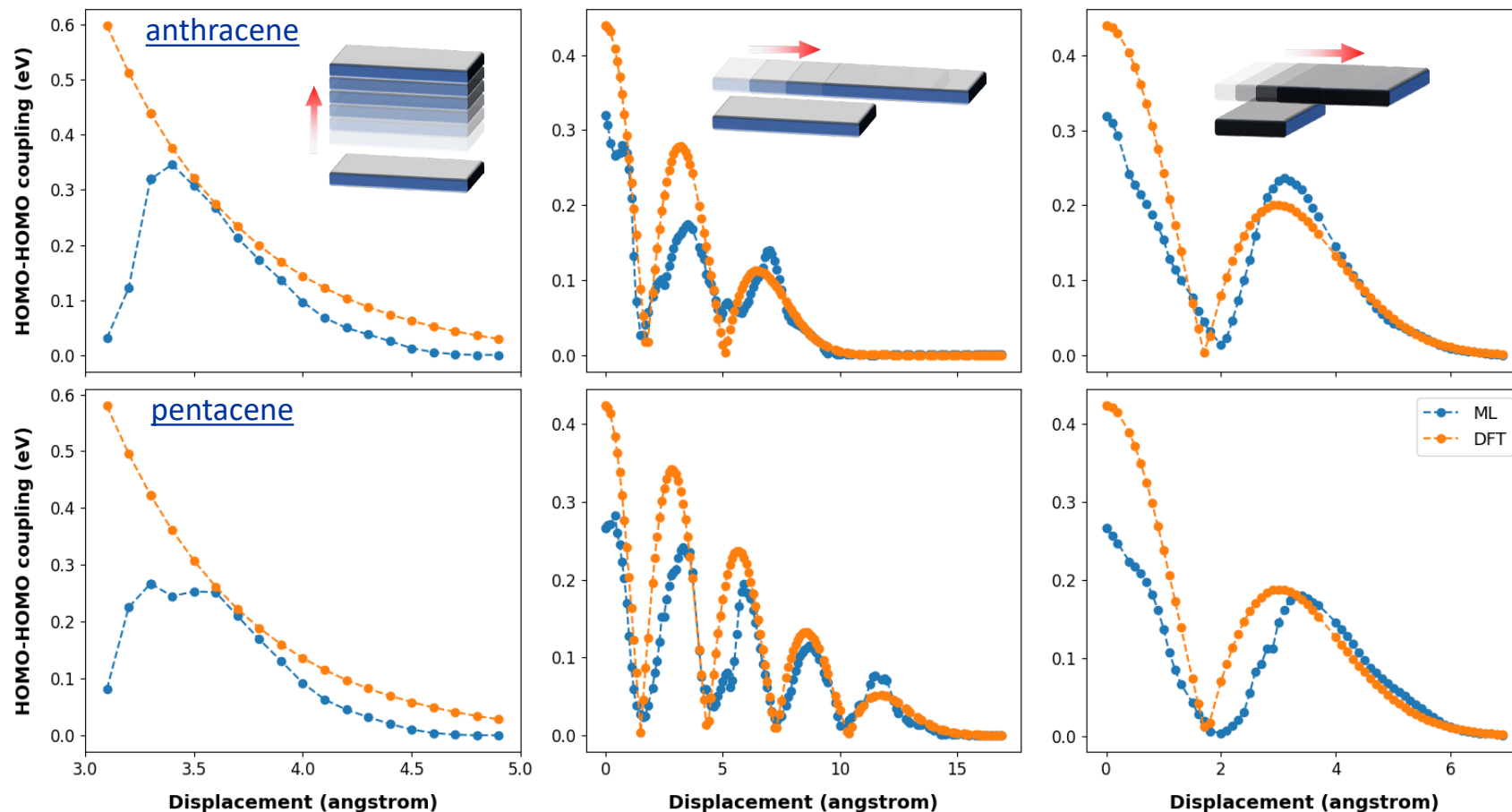
Select property to predict

Adiabatic Ionization Energy (R2 = 0.87; MAE = 0.16)

Predict

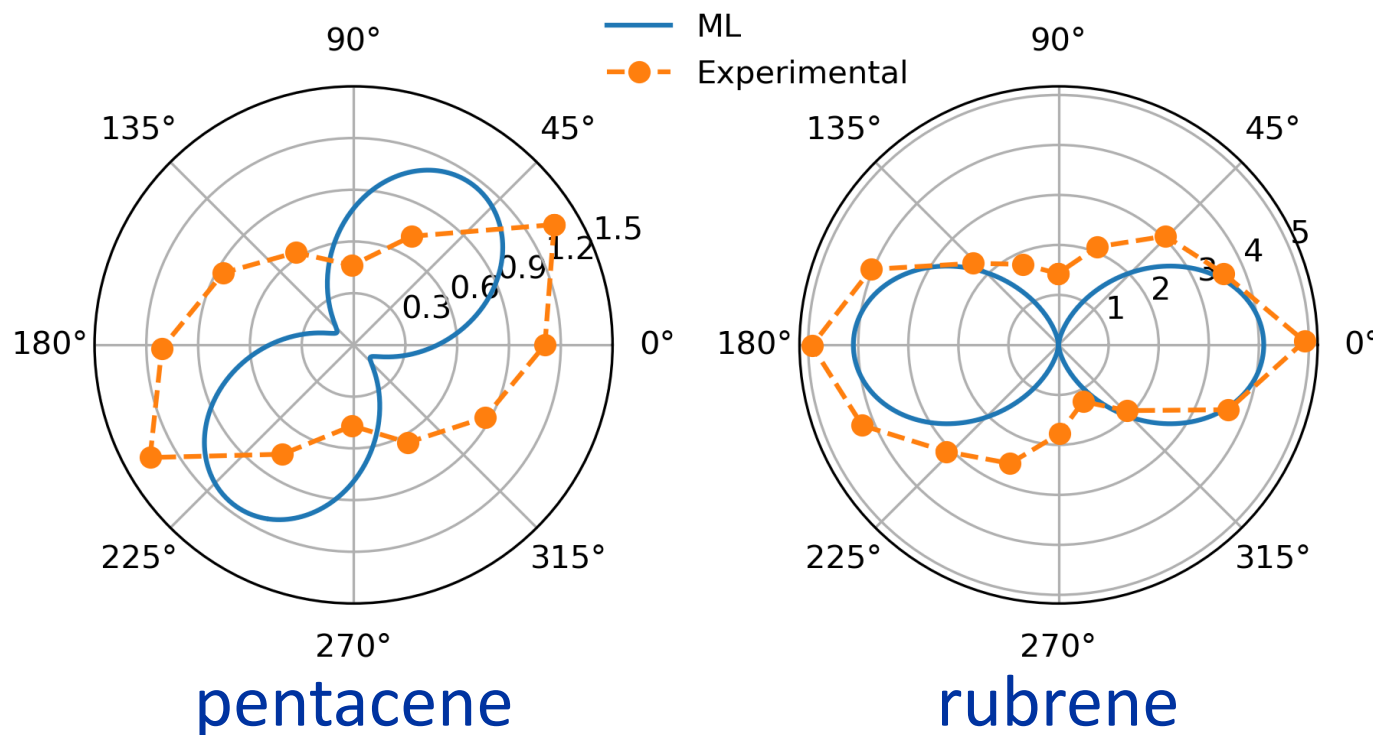
Predicted value is : **7.16 ± 0.151 eV**

- OCLEOT dimer v1 dataset has >439,000 DFT-determined electronic couplings



via semi-empirical Marcus theory, prediction of charge-carrier mobility anisotropy

- ML intramolecular reorganization energy
- ML intermolecular electronic couplings



OCELOT ML



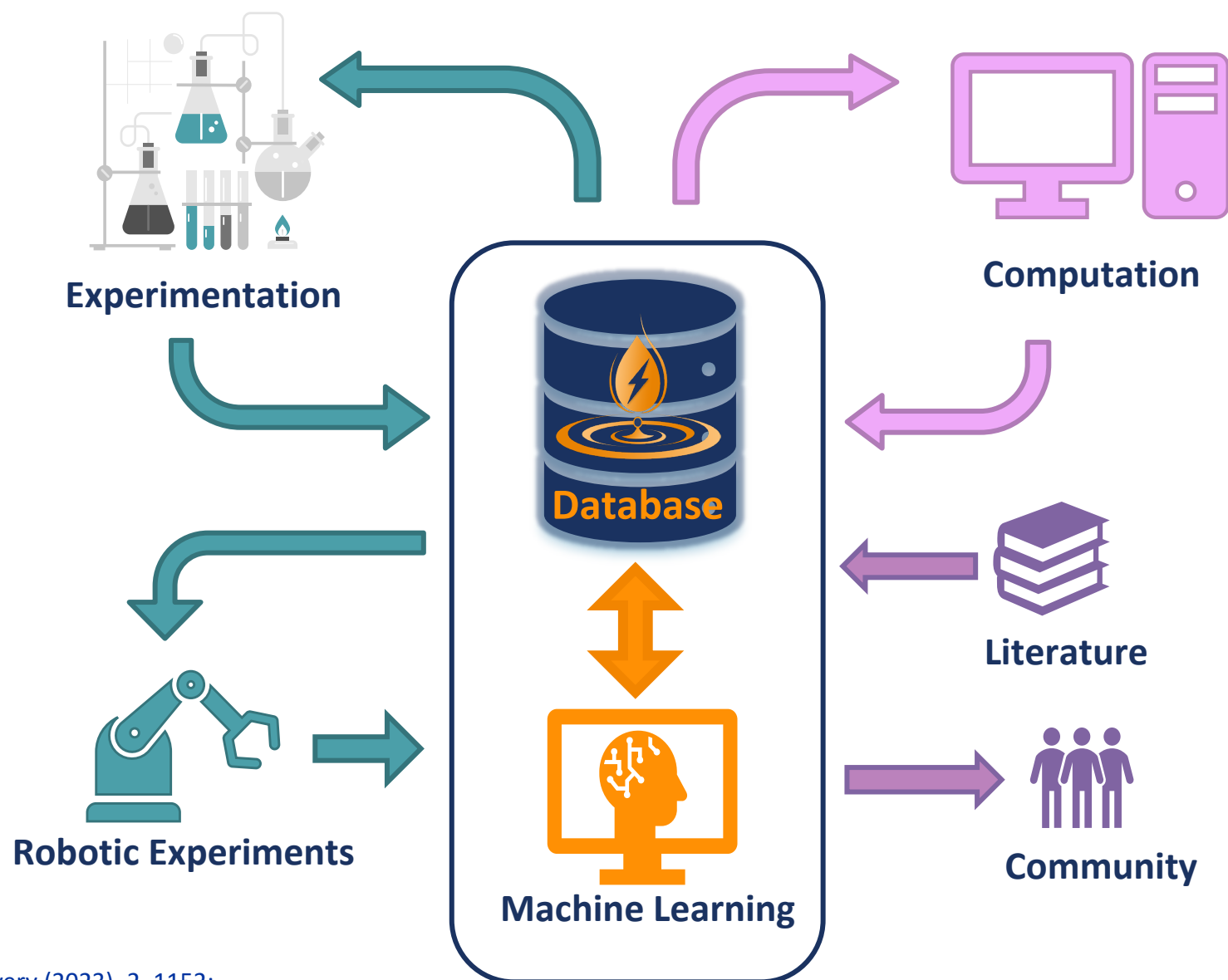
oscar.as.uky.edu/ocelotml



Data-enabled Discovery and Design to Transform Liquid-based Energy Storage (D³TaLES)



d3tales.as.uky.edu



How can we incorporate (dynamic) **experimental data**?

How do we collect all the data associated with an experiment, i.e., the **metadata**?

A chemist performs an experiment,
hand-writing measurements and
observations in a laboratory notebook...

1

Missing details

Only selected details are transferred to the published paper. Other chemists have difficulty reproducing results.

2

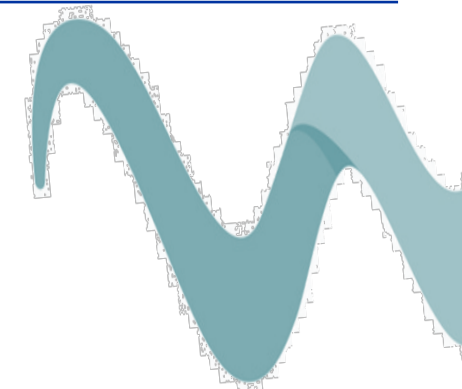
Machine readability

It is difficult for computers to translate notebook text to experiment steps for analysis

3

Manual analysis

The chemist must manually transfer values from notebook to computer, then perform calculations.



ExpFlow rdu230@uky.edu

Workflow:

Sequence	Name	Description	Start Position	End Position	Edit
1	transfer_solid	Add redox molecule	Quinone	Mixing Beaker	Edit

Calculate Meta Properties

Charge:

Diffusion Coefficient	Charge Transfer Rate	Oxidation Potential (Avg. $E_{1/2}$)
1.465e-05 cm ² /s (mean method: 1.973e-05 cm ² /s)	6.916e-03 cm/s	0.3498 V
<input type="button" value="Approve Data"/>	<input type="button" value="Approve Data"/>	<input type="button" value="Approve Data"/>

Customizable

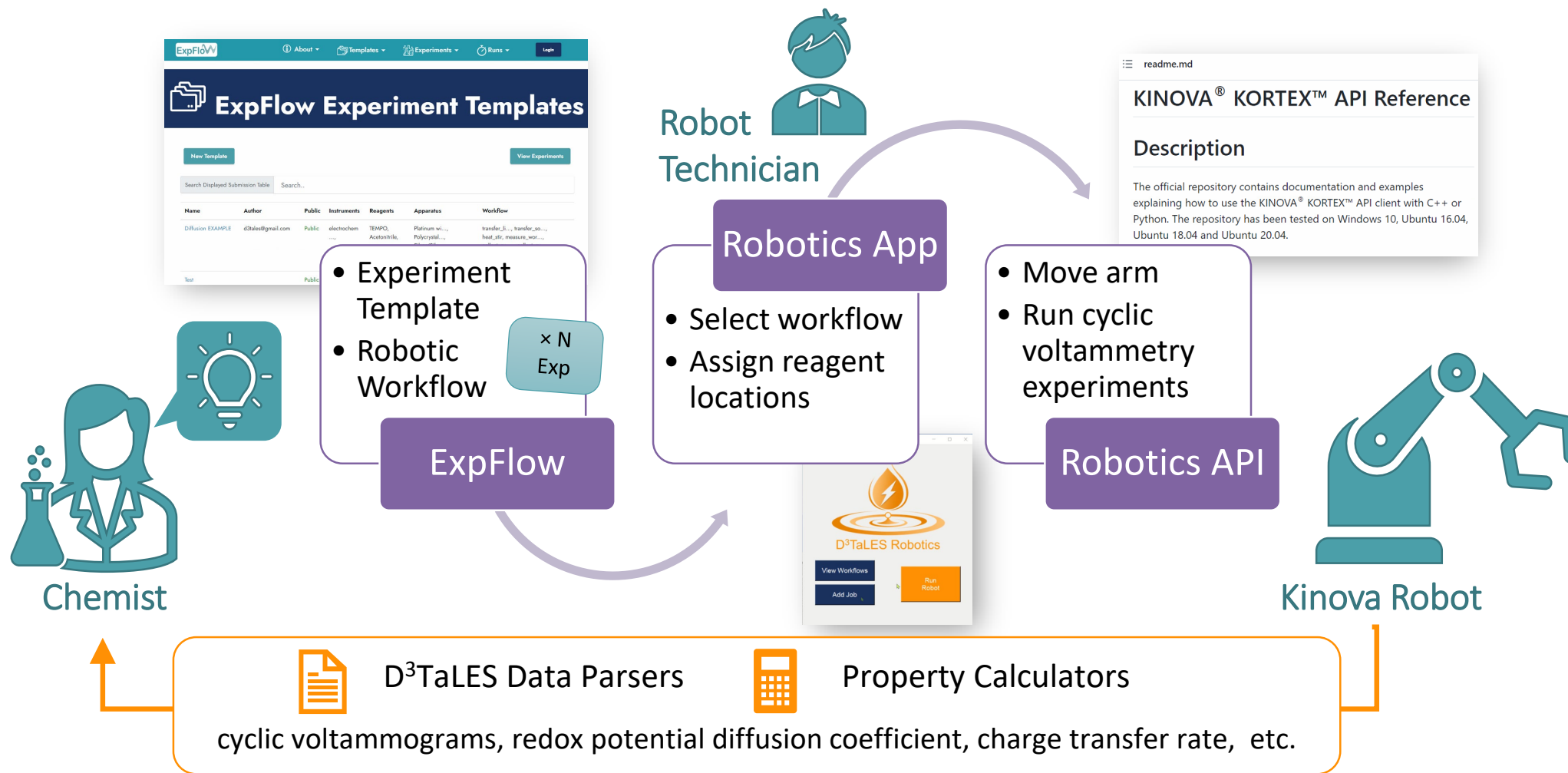
Users can modify the tools for specific use

Reusable

Experimental templates can be used by others

Sharable

All raw data and experimental procedures can be shared

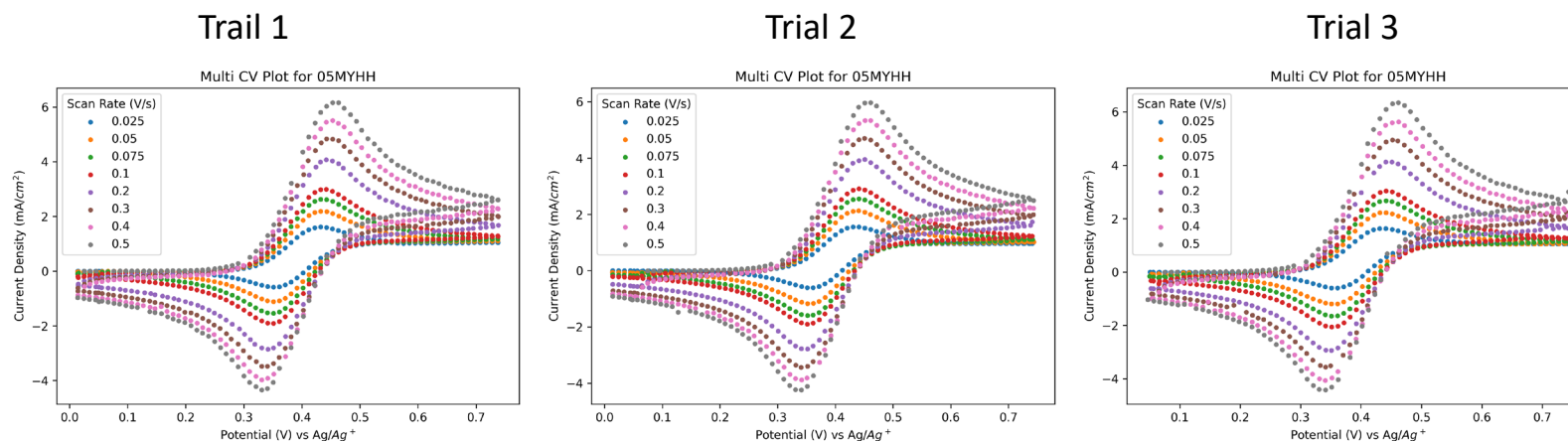




R. Duke, V. Bhat, P. Sornberger, S.A. Odom & C. Risko, *Digital Discovery* (2023), 2, 1152;

R. Duke, S. Mahmoudi, A.P. Kaur, V. Bhat, I. Dingle, N.C. Stumme, S.K. Shaw, D. Eaton, A. Vego & C. Risko (2023), submitted.

Exp 2: N-[2-(2-Methoxyethoxy)ethyl]-phenothiazine (MEEPT) CVs



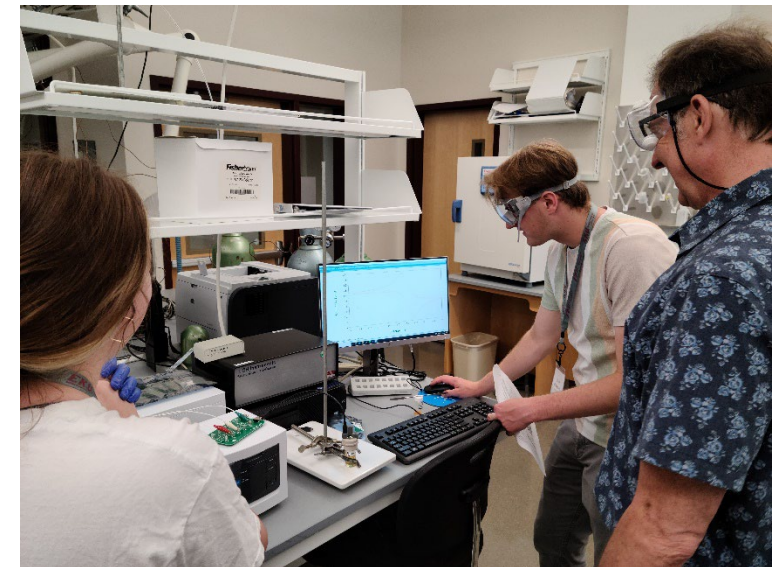
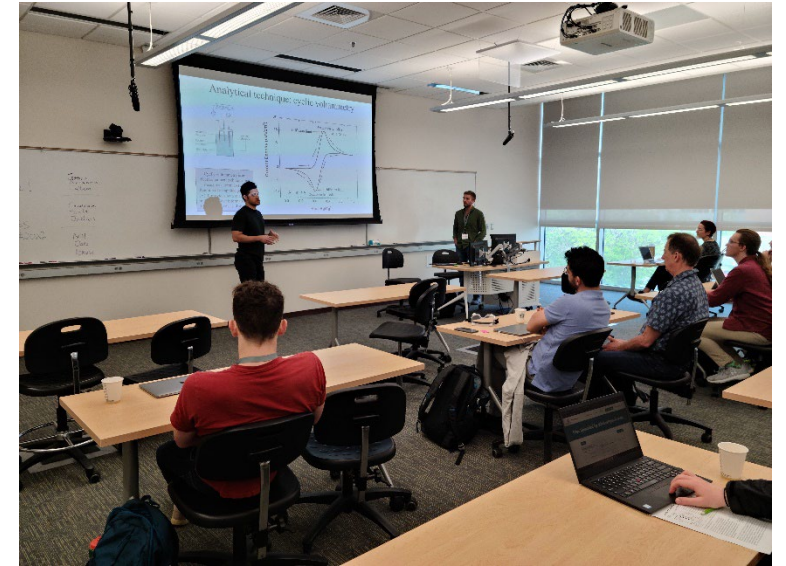
ROM	Robotic/ExpFlow						Literature Reported vs. Ag/Ag+		Literature Reported vs. Fc/Fc+	
	Trial 1	Trial 2	Trial 3	Avg. (vs. Ag/Ag ⁺)	Std. Dev.	vs. Fc/Fc ⁺ α	Value	Ref.	Value	Ref.
Fc	0.081	0.082	0.082	0.082	0.001	0.000	0.086	Ref. ⁶		
MEEPT	0.396	0.396	0.396	0.396	0.000	0.314	0.410*	Ref. ¹⁴	0.310	Ref. ⁸
DMPZ	-0.156	-0.156	-0.156	-0.156	0.000	-0.238	-0.150	Ref. ⁹		
4-MeO TEMPO	0.371	0.376	0.375	0.374	0.003	0.292	0.68 ⁺⁺	Ref. ⁵		
DBB	0.773	0.773	0.773	0.773	0.000	0.691	0.710	Ref. ¹⁰		
DBBB	0.773	0.768	0.773	0.771	0.003	0.690			0.60 [†]	Ref. ¹³
TH	0.910	0.910	0.910	0.910	0.000	0.828	0.900	Ref. ¹²	0.840	Ref. ⁷
ECZ**	0.678	0.678	0.672	0.676	0.003	0.594				

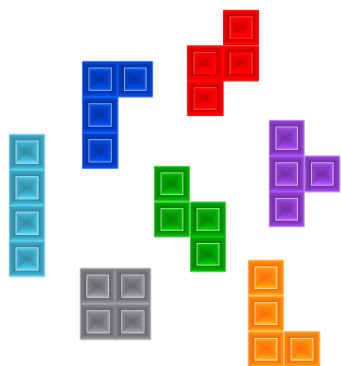
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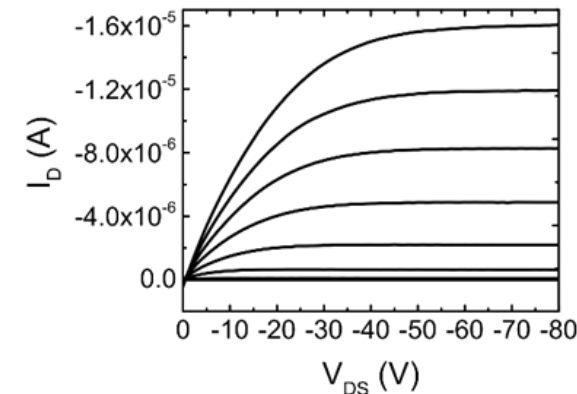


In the Mix: An immersive educational experience





A library of organic chromophores



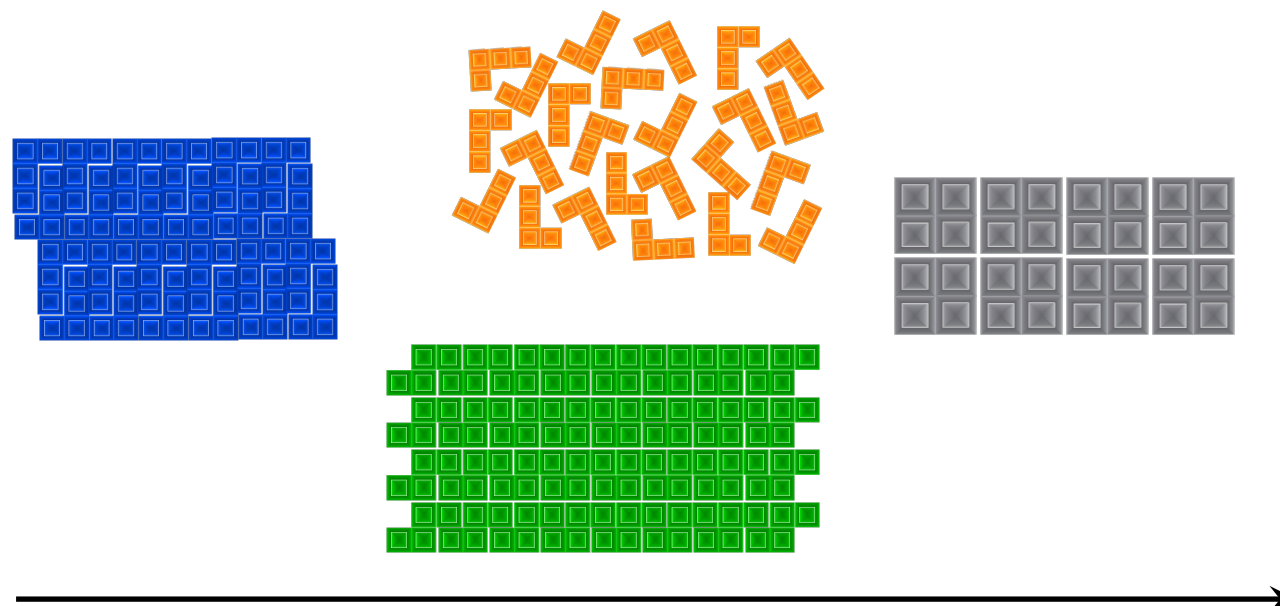
A. Zeidell *et al.* Chem. Mater. (2019) 31, 6962.

BUILDING BLOCK

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THANK YOU!